

Cycloaddition Behavior of 2-Substituted Norbornadienes towards 4-Phenyl-4*H*-1,2,4-triazole-3,5-dione (PTAD): Homo Diels-Alder Reactivity versus Insertion, Rearrangement, and [2 + 2] Cycloaddition

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The reaction of PTAD with 2-trimethylsilyl-, 2-chloro-, 2-cyano-, and 2-methoxycarbonylnorbornadienes **1a–d**, respectively, was investigated. In all cases homo Diels-Alder adducts were obtained, for **1a** the unexpected regioisomer, the 1-trimethylsilyl derivative **2a**, and for other norbornadienes **1b–d** the usual cyclopropane-substituted products **2b–d**. Except for the trimethylsilyl system **1a**, for which also the dicarboximides **4a** and **4'a**, respectively, (rearrangement urazoles) were obtained, the other norbornadienes **1b–d** afforded the insertion products **5b–d**. With increasing electron-withdrawal by the 2-substituents the insertion products **5b–d** increased at the expense of the homo Diels-Alder adducts **2b–d**. These results are mechanistically rationalized in terms of stepwise cycloaddition via 1,5-dipolar intermediates. In the case of 2-chloronorbornadiene (**1b**), besides the homo Diels-Alder **2b**, the rearrangement urazole **4b** and the insertion product **5b**, also the [2 + 2] cycloadduct **3b** was formed.

Cycloadditionsverhalten von 2-substituierten Norbornadienen mit 4-Phenyl-4*H*-1,2,4-triazol-3,5-dion (PTAD): Homo-Diels-Alder-Reaktivität gegenüber Einschiebung, Umlagerung und [2 + 2]-Cycloaddition

Die Reaktion von PTAD mit den 2-Trimethylsilyl-, 2-Chlor-, 2-Cyan- oder 2-Methoxycarbonylnorbornadienen **1a–d** wurde untersucht. In allen Fällen wurden Homo-Diels-Alder-Addukte erhalten, von **1a** das unerwartete Regiosomere 1-Trimethylsilylderivat **2a** und von **1b–d** die gewöhnlichen cyclopropansubstituierten Produkte **2b–d**. Zusätzlich führte das Trimethylsilylsystem **1a** zu den Dicarboximiden **4a** bzw. **4'a** (Umlagerungsurazole). Die Norbornadiene **1b–d** bildeten die Insertionsprodukte **5b–d**. Mit steigendem Elektronenzug der 2-Substituenten werden die Einschiebungsprodukte **5b–d** auf Kosten der Homo-Diels-Alder-Addukte **2b–d** in steigendem Maß gebildet. Diese Ergebnisse erklären wir unter der Annahme eines mehrstufigen Cycloadditionsmechanismus über 1,5-dipolare Zwischenstufen. Im Fall von 2-Chlornorbornadien (**1b**) wurde neben dem Homo-Diels-Alder-Addukt **2b**, dem Umlagerungsurazol **4b** und dem Einschiebungsprodukt **5b** auch das [2 + 2]-Cycloaddukt **3b** erhalten.

The usual cycloaddition of dienophiles with norbornadienes is the homo Diels-Alder reaction¹⁾. In the case of carbenes and singlet oxygen, such electrophilic species lead besides homo Diels-Alder products also to [2 + 2] cycloadducts²⁾. In fact, with difluorocarbene [2 + 2] cycloaddition predominates; but with increasing electron-withdrawing substituents at the 2-position of the norbornadiene, the proportion of homo Diels-Alder reaction decreases. This interesting reactivity pattern has been rationalized³⁾ in terms of LUMO carbene – HOMO diene interaction and polarization of the π bonds by the 2-substituents, suggesting that other electrophilic dienophiles should portray this cycloaddition behavior.

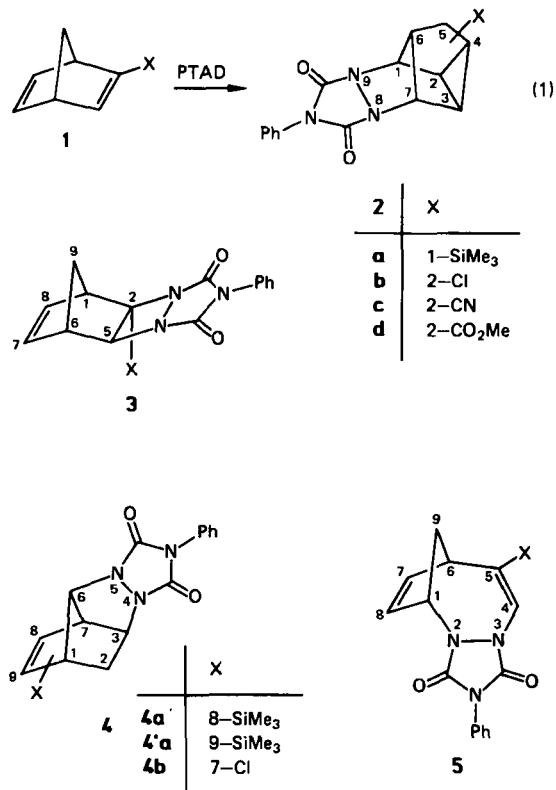
Previously we reported³⁾ our preliminary results on the reaction of PTAD with 2-chloronorbornadiene, which exhibits most unusual cycloaddition behavior (Eq. 1; X = Cl). While precedents for homo Diels-Alder, [2 + 2], and rearrangement products **2**, **3**, and **4** (X = Cl), respectively, are documented⁴⁾, the unusual insertion product **5** (X = Cl) was new.

Since in the difluorocarbene reaction 2-substituents in the norbornadiene markedly influence the cycloaddition mode (homo Diels-Alder versus [2 + 1] reactivity)²⁾, it was our interest to explore such substituent effects on the product distribution for the PTAD reaction. The results are sum-

marized in Table 1. Before entering into the discussion of these results, it should be mentioned that also 2-methoxy- and 2-(trimethylsilyloxy)norbornadienes were examined, but both led to intractable and complex reaction mixtures and work on these derivatives was abandoned.

For comparison purposes, the data of the previously reported³⁾ 2-chloronorbornadiene (**1b**) are also included in Table 1. Clearly, this substrate exhibits the most diversified and complex product pattern of the 2-substituted norbornadienes examined here. Moreover, the poor product balance (ca. 30%; Table 1) must be kept in mind. Most of the 2-chloronorbornadienes lead to undefined, high-molecular-weight material, which is retained on the silica gel column during chromatography of the crude reaction mixture. While the spectral data, especially ^1H - and ^{13}C -NMR and ^1H -NMR decoupling experiments, permitted unequivocal characterization of the cycloadducts **2b**, **3b**, and **4b** (cf. Experimental), an X-ray analysis was essential in determining the structure of the unprecedented insertion product **5b**⁵⁾. The structures of the remaining insertion products **5c** and **5d**, as well as the other cycloadducts, were arrived at by

comparison of the spectral data with those of the 2-chloronorbornadiene products (cf. Experimental).



The product composition in Table 1 for the 2-trimethylsilyl-, 2-chloro-, 2-cyano- and 2-methoxycarbonylnorbornadienes **1a–d** is mechanistically perplexing. Thus, all norbornadienes give the homo Diels-Alder adducts **2a–d**. However, while the 2-chloro-, 2-cyano- and 2-methoxycarbonyl substituents lead to the expected^{2b)} regioisomers **2b–d** with the substituents on the cyclopropane ring, the 2-trimethylsilyl group gives the other regioisomeric product **2a**. Moreover, the trimethylsilyl system **1a** additionally affords the rearrangement urazoles **4a** and **4'a** (1:1 mixture of 8- and 9-trimethylsilyl regioisomers), the cyano and methoxycarbonyl derivatives **1c** and **1d**, respectively, the inser-

tion products **5c** and **5d**, and only the 2-chloronorbornadiene (**1b**) yields both the rearrangement urazole **4b** and the insertion product **5b**, together with the [2 + 2] adduct **3b**. For the norbornadienes **1a**, **1c**, and **1d** the cycloadditions proceed with more than 95% product balance (Table 1), but **1b** leads predominantly to undefined high-molecular-weight material. Furthermore, the qualitative reactivity order is that the silyl derivative reacts faster than the cyano and methoxycarbonyl cases, as one would expect for the electrophilic PTAD⁶⁾.

In terms of frontier molecular orbital theory^{2b)} it is difficult to rationalize the product data of Table 1 for the cycloaddition of PTAD and the norbornadienes **1**. The only trend of the difluorocarbene reactivity pattern that is paralleled by PTAD is that with increasing electron-withdrawing character of the 2-substituent the degree of homo Diels-Alder cycloaddition is diminished. While for the difluorocarbene reaction the diminution of the homo Diels-Alder product **2** is offset by an increase in the [2 + 2] cycloadduct **3**²⁾, for PTAD the amount of insertion product **5** augments, except for the 2-trimethylsilyl system **1a**, which instead generates the rearrangement product **4a**.

These divergent reaction paths complicate of course the mechanistic interpretation. However, interesting is the fact that the electron-withdrawing substituents 2-chloro, 2-cyano, and 2-methoxycarbonyl give with PTAD cyclopropane-substituted regioisomers of the homo Diels-Alder adducts **2b–d**, while the electron-donating 2-trimethylsilyl group leads to the other regioisomer **2a**.

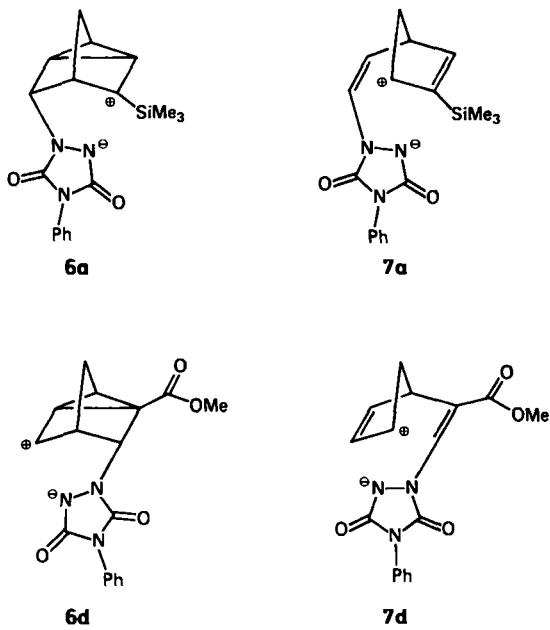
Assuming that the PTAD reaction with these substituted norbornadienes proceeds stepwise through dipolar ions^{2b)}, for the 2-trimethylsilyl case the 1,5-dipole **6a** should be preferred in view of additional stabilization of the positive pole by the α-trimethylsilyl group. This stabilized cyclopropylcarbinyl cation would have less incentive to open the ring to give the dipole **7a**, a potential precursor of a regioisomeric insertion product of **5a**. Consequently, cyclization of the dipole **6a** leads to the homo Diels-Alder adduct and not to an insertion product. On the other hand, taking the 2-methoxycarbonyl case as example, the preferred 1,5-dipole should be **6d**, with the electron-withdrawing group as remote from the positive pole as is feasible. Cyclization of dipole **6d** to the homo Diels-Alder adduct **2d** is competed by ring-ope-

Table 1. Product composition of the cycloaddition of the norbornadienes **1a–d** and PTAD

Norbornadiene	Reaction time [h] ^{a)}	Product balance (%) ^{b)}	Homo Diels-Alder 2	Relative product yields (%) ^{c)}		
				[2 + 2] 3	Rearrangement 4	Insertion 5
1a (2-SiMe ₃)	36	>95	77 ^{d)}	—	23 ^{e)}	—
1b (2-Cl)	48	29 ^{f)}	73	12	4.5	10.5
1c (2-CN)	96	>95	67	—	—	33
1d (2-CO ₂ Me)	48	>95	41	—	—	59

^{a)} For complete consumption of **1**, addition of excess PTAD is required. — ^{b)} Total yield of isolated product by gravimetry. — ^{c)} Obtained by quantitative ¹H-NMR (400 MHz) analysis of the crude reaction mixture prior to workup; normalized to 100%; values accurate within ca. 5% of stated values. — ^{d)} Instead of the expected 2-trimethylsilyl, the 1-trimethylsilyl regioisomer was formed. — ^{e)} A 1:1 mixture of 8- and 9-trimethylsilyl derivatives was obtained. — ^{f)} Large quantities of intractable high molecular weight material were formed.

ning to the stabilized dipole **7d** (conjugated α,β -unsaturated ester) and subsequent collapse to the corresponding insertion product **5d**.



It would have been important to substantiate these mechanistic arguments by employing 2-phenyl- and 2-benzenesulfonyl-substituted norbornadienes. Unfortunately, the latter does not react with PTAD and the former undergoes preferentially Diels-Alder cycloadditions with the styryl moiety.

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Experimental

IR spectra were run on the Perkin-Elmer Models 1420 and 157G. — $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ data were obtained with the following spectrometers: 60 and 22.6 MHz, Hitachi-Perkin-Elmer R-24 B; 90 MHz, Varian EM 390; 200 and 50.3 MHz, Bruker AS-200; and 400 and 100 MHz, Bruker WM-400. Chemical shifts (δ values) are given relative to tetramethylsilane for protons and deuteriochloroform for carbons. — Melting points were taken with a Reichert ThermoVar Kofler apparatus and are uncorrected. — Combustion analyses for elemental composition were run in-house. — For thin-layer chromatography (TLC) Polygram SIL/G/UV (40 \times 80 mm, Macherey, Nagel & Co.) and for column chromatography silica gel (70–230 mesh or 32–63 μm , activity grade I, Merck) were employed.

Commercial reagents and solvents were purified to match the reported physical and spectral data. Unless otherwise stated, rotovaporation was carried out at 20–25°C (room temperature, abbrev. R. T.) and 10–20 Torr (water aspirator).

2-(Trimethylsilyl)bicyclo[2.2.1]hepta-2,5-diene⁷ (**1a**): Starting from 3.69 g (40.0 mmol) of norbornadiene in 10 ml of dry THF,

2.47 g (22.0 mmol) of tBuOK in 20 ml of dry THF, 17 ml of 1.3 M (22.1 mmol) n-BuLi in hexane, and 3.04 g (28.0 mmol) of trimethylsilyl chloride in 5 ml of THF was obtained 2.60 g (72%) of **1a** as colorless liquid, b.p. 58–60°C at 18 Torr (ref.⁷) 53°C at 15 Torr). — $^1\text{H NMR}$ (CDCl_3 , 200 MHz): δ = 0.08 (s; 9H, CH_3), 1.90 (d, $J_{7s,7a}$ = 5.9 Hz; 2H, 7s-H and 7a-H), 3.66 (m; 1H, 1-H or 4-H), 3.73 (m; 1H, 1-H or 4-H), 6.70 (m; 2H, 5-H and 6-H), 7.04 (d, $J_{3,4}$ = 2.7 Hz; 1H, 3-H). — $^{13}\text{C NMR}$ (50.3 MHz, CDCl_3): δ = −2.08 (q; CH_3), 51.97 (d), 53.32 (d), 74.06 (t; C-7), 142.33 (d), 143.16 (d), 153.29 (d; C-3), 155.72 (s; C-2).

2-Chlorobicyclo[2.2.1]hepta-2,5-diene¹¹ (**1b**): From 20.0 g (0.124 mol) of 5,6-dichloronorbornene and 10.0 g (0.178 mol) of potassium hydroxide in 40 ml of ethylene glycol was obtained after distillation 5.60 g (36%) of **1b**, b.p. 94–96°C at 14 Torr (ref.¹¹) 140°C at 769 Torr), n_D^{22} = 1.4960. — The product polymerizes on standing at R. T. — IR (CCl_4): 3040 cm^{-1} , 2980, 2950, 2875, 1585, 1550, 1450, 1420, 1335, 1300, 1245, 1150, 1095, 1035, 1005, 910, 860, 835, 700. — $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ = 2.10 (dt, $J_{7s,7a}$ = 6.0 Hz, $J_{1,7s}$ = $J_{4,7s}$ = 1.5 Hz; 1H, 7s-H), 2.26 (dt, $J_{7s,7a}$ = 6.0 Hz; $J_{1,7a}$ = $J_{4,7a}$ = 1.5 Hz; 1H, 7a-H), 3.37 (m; 1H, 4-H), 3.61 (dm, $J_{1,6}$ = 2.6 Hz; 1H, 1-H), 6.39 (d, $J_{3,4}$ = 3.3 Hz; 1H, 3-H), 6.79 (dd, $J_{5,6}$ = 5.0 Hz, $J_{1,6}$ = 2.8 Hz; 1H, 6-H), 6.89 (dd, $J_{5,6}$ = 5.0 Hz, $J_{4,5}$ = 3.0 Hz; 1H, 5-H).

Bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile⁸ (**1c**): To a solution of cyanoacetylene⁹ (2.80 g, 54.9 mmol) in 30 ml of methylene chloride, was added monomeric cyclopentadiene (5.5 ml, 66.8 mmol) at 0°C while stirring. The mixture was stirred for about 12 h at R. T. and the solvent roto-evaporated. The residue was purified by silica gel chromatography [1:30 substrate/adsorbent ratio, 2:8 methylene chloride/petroleum ether (30–50) as eluent], resulting in 5.70 g (90%) of **1c** as colorless liquid. The product decomposes at R. T. — IR (CCl_4): 3080 cm^{-1} , 3010, 2990, 2950, 2875, 2210, 1750, 1580, 1450, 1230, 1210, 1175, 1015, 920, 910, 885, 875, 855, 710, 610. — $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ = 2.16 (dt, $J_{7s,7a}$ = 6.7 Hz, $J_{1,7s}$ = $J_{4,7s}$ = 1.2 Hz; 1H, 7s-H), 2.19 (dt, $J_{7s,7a}$ = 6.7 Hz, $J_{1,7a}$ = $J_{4,7a}$ = 1.5 Hz; 1H, 7a-H), 3.84 (m; 2H, 1-H and 4-H), 6.75 (ddd, $J_{5,6}$ = 5.0 Hz, $J_{4,5}$ = 3.2 Hz, $J_{3,5}$ = 0.6 Hz; 1H, 5-H), 6.90 (dd, $J_{5,6}$ = 5.6 Hz, $J_{1,6}$ = 3.0 Hz; 1H, 6-H), 7.66 (d, $J_{3,4}$ = 3.1 Hz; 1H, 3-H). — $^{13}\text{C NMR}$ (CDCl_3 , 100 MHz): δ = 51.77 (d), 53.58 (d), 74.75 (t; C-7), 117.04 (s; C-2), 141.60 (d), 142.75 (d), 161.17 (d), 193.29 (s; CN). — MS (70 eV): m/z (%) = 118 (8, $\text{M}^+ + 1$), 117 (100, M^+), 116 (88), 104 (11), 91 (66), 90 (65), 89 (41), 77 (13), 66 (98, C_6H_5^+), 51 (24). [The m/z = 117 peak overlaps with ca. 8% $^{13}\text{C}^{12}\text{C}_2\text{H}_7\text{N}$ ($\text{M} - 1$)]

$\text{C}_8\text{H}_7\text{N}$ Calcd. 117.0579 Found 117.0575

Methyl Bicyclo[2.2.1]hepta-2,5-diene-2-carboxylate¹⁰ (**1d**): A 25-ml three-necked flask provided with nitrogen inlet, outlet tubes, and a magnetic spinbar was flame-dried under a stream of nitrogen and charged with methyl acetylenecarboxylate (2.00 g, 23.8 mmol) and freshly distilled cyclopentadiene (1.57 g, 23.8 mmol). The reaction mixture was allowed to stir for about 12 h at 50°C, and distillation gave 2.50 g (70%) of **1d**, b.p. 80–85°C at 12 Torr (ref.¹⁰) 28°C at 0.1 Torr). — $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ = 2.14 (d, $J_{7s,7a}$ = 6.0 Hz; 2H, 7-H), 3.75 (br. s; 4H, CH_3 and 4-H), 6.76 (dd, $J_{5,6}$ = 5.1 Hz, J = 3.1 Hz; 1H, 5-H or 6-H), 6.93 (dd, $J_{5,6}$ = 5.1 Hz, J = 3.1 Hz; 1H, 6-H or 5-H), 7.71 (d, $J_{3,4}$ = 3.2 Hz; 1H, 3-H). — $^{13}\text{C NMR}$ (CDCl_3 , 100 MHz): δ = 49.73 (d), 50.19 (q; CH_3), 51.40 (d), 74.33 (t; C-7), 141.66 (d), 143.52 (d), 148.95 (s; C-2), 156.31 (d), 165.50 (s; C=O).

General Procedure for the Cycloaddition of PTAD to the Norbornadienes 1: To a solution of the particular norbornadiene **1** in CH_2Cl_2 was added at once, while stirring magnetically, at 0°C a stoichiometric amount of PTAD. The reaction mixture was allowed to stir at R. T. The reaction progress was monitored by TLC (silica

gel, CH_2Cl_2 as eluent). The reactive norbornadienes **1a** and **1b** were consumed within 48–72 h, while for the less reactive **1c** and **1d** the PTAD was in part consumed via decomposition. Accordingly, each day another ca. 0.50 g of PTAD was added until all of the norbornadiene had been consumed. The dark-brown reaction mixture was concentrated by roto-evaporation and the residue submitted to silica gel chromatography (1:30 substrate/adsorbent ratio), affording the corresponding cycloadducts. Final purification entailed recrystallization. The experimental details for each particular case are given below.

2-(Trimethylsilyl)norbornadiene (1a**):** From 1.00 g (6.09 mmol) of **1a** and 1.02 g (6.11 mmol) of PTAD in 50 ml of CH_2Cl_2 were obtained the cycloadducts **2a**, **4a**, and **4'a** after 36 h of reaction time and silica gel chromatography using petroleum ether (30–50)/ethyl acetate as eluent. **4a** and **4'a** were eluted as first fraction and separated by fractional crystallization using AcOEt as solvent.

N-Phenyl-8-(trimethylsilyl)-4,5-diazatricyclo[4.3.0.0^{3,7}]non-8-ene-4,5-dicarboximide (4a**):** 310 mg (15%) of colorless plates, m.p. 160–161°C (AcOEt). — IR (KBr): 3035 cm⁻¹, 3020, 2960, 2950, 1720, 1655, 1595, 1410, 1320, 1290, 1260, 1250, 1240, 1130, 1090, 875, 840, 830, 770, 760, 725, 690, 645. — ¹H NMR (400 MHz, CDCl_3): δ = 0.10 (s; 9 H, CH_3), 1.31 (ddd, $J_{2n,2x} = 12.6$ Hz, $J_{3,2n} = 5.0$ Hz, $J_{2n,1} = 1.5$ Hz; 1 H, 2n-H), 1.90 (dd, $J_{2n,2x} = 12.6$ Hz, $J_{2x,1} = 5.1$ Hz; 1 H, 2x-H), 3.23 (m; 2 H, 1-H and 7-H), 4.35 (dd, $J_{3,2n} = 5.0$ Hz, $J_{3,6} = 1.9$ Hz; 1 H, 3-H), 4.38 (m; 1 H, 6-H), 6.59 (d, $J = 3.8$ Hz; 1 H, 9-H), 7.34–7.54 (m; 5 H, Ph). — ¹³C NMR (100 MHz, CDCl_3): δ = -2.12 (q; CH_3), 33.39 (t; C-2), 46.05 (d), 54.98 (d), 57.07 (d), 77.19 (d), 125.44 (d), 128.25 (d), 129.14 (d), 141.67 (s), 148.44 (d), 156.07 (s; C=O), 156.35 (s; C=O). — MS (70 eV): m/z (%) = 340 (8, $\text{M}^+ + 1$), 339 (29, M^+), 324 (3), 220 (4), 192 (7), 163 (4), 119 (17), 105 (18), 100 (23), 91 (10), 78 (19), 73 [100, $\text{Si}(\text{CH}_3)_3^+$], 59 (10).

$\text{C}_{18}\text{H}_{21}\text{N}_3\text{O}_2\text{Si}$ (339.5) Calcd. C 63.69 H 6.24 N 12.38
Found C 63.93 H 6.28 N 12.53

N-Phenyl-9-(trimethylsilyl)-4,5-diazatricyclo[4.3.0.0^{3,7}]non-8-ene-4,5-dibarboximide (4'a**):** 150 mg (8%), colorless powder, m.p. 146–148°C (AcOEt). — IR (KBr): 3100 cm⁻¹, 3000, 2960, 1800, 1750, 1510, 1440, 1330, 1300, 1260, 1250, 1140, 1095, 870, 845, 780, 760, 715, 700, 640. — ¹H NMR (400 MHz, CDCl_3): δ = 0.11 (s; 9 H, CH_3), 1.19 (ddd, $J_{2n,2x} = 12.6$ Hz, $J_{2n,3} = 5.0$ Hz, $J_{2n,1} = 1.5$ Hz; 1 H, 2n-H), 1.88 (dd, $J_{2n,2x} = 12.6$ Hz, $J_{1,2x} = 4.9$ Hz; 1 H, 2x-H), 3.23 (m; 1 H, 7-H), 3.28 (m; 1 H, 1-H), 4.38 (m; 1 H, 6-H), 4.43 (dd, $J_{2n,3} = 4.9$ Hz, $J_{3,7} = 2.0$ Hz; 1 H, 3-H), 6.13 (dd, $J_{7,8} = 3.0$ Hz, $J_{1,8} = 1.1$ Hz; 1 H, 8-H), 7.32–7.56 (m; 5 H, Ph). — ¹³C NMR (100 MHz, CDCl_3): δ = -2.11 (q; CH_3), 33.11 (t; C-2), 47.21 (d), 53.59 (d), 57.32 (d), 77.76 (d), 125.43 (d), 128.24 (d), 129.14 (d), 131.75 (s), 135.21 (d), 148.44 (d), 155.53 (s; C=O), 156.17 (s; C=O). — MS (70 eV): m/z (%) = 341 (1, $\text{M}^+ + 2$), 340 (6, $\text{M}^+ + 1$), 340 (26, M^+), 220 (3), 219 (2), 205 (3), 192 (6), 178 (3), 163 (3), 135 (7), 132 (5), 119 (16), 105 (18), 100 (24), 91 (11), 78 (20), 59 (10).

$\text{C}_{18}\text{H}_{21}\text{N}_3\text{O}_2\text{Si}$ (339.5) Calcd. C 63.69 H 6.24 N 12.38
Found C 63.93 H 6.28 N 12.53

1-(Trimethylsilyl)-N-phenyl-8,9-diazatetracyclo[4.3.0.0^{2,4,0^{3,7}]nonane-8,9-dicarboximide (2a**):}** was isolated as second fraction, 1.36 g (66%), colorless plates, m.p. 155–156°C (AcOEt). — IR (KBr): 3080 cm⁻¹, 3000, 2960, 2890, 1760, 1700, 1600, 1595, 1505, 1490, 1455, 1400, 1325, 1270, 1245, 1215, 1135, 1125, 1110, 1085, 1075, 1025, 935, 895, 855, 840, 800, 770, 745, 700, 690, 660, 645, 630. — ¹H NMR (400 MHz, CDCl_3): δ = 0.29 (s; 9 H, CH_3), 1.52–1.64 (m; 3 H, 2-H, 3-H and 4-H), 1.76 (dt, $J_{5s,5a} = 11.8$ Hz, $J = 1.3$ Hz; 1 H, 5-H), 1.83 (dt, $J_{5s,5a} = 11.8$ Hz, $J = 1.4$ Hz; 1 H, 5-H), 2.56 (m; 1 H, 6-H), 4.46 (t, $J = 2.1$ Hz; 1 H, 7-H), 7.31–7.54

(m; 5 H, Ph). — ¹³C NMR (100 MHz, CDCl_3): δ = -1.94 (q; CH_3), 13.18 (d), 15.85 (d), 28.81 (t; C-5), 46.70 (d), 65.60 (d; C-7), 67.51 (s; C-1), 125.41 (d), 127.89 (d), 132.09 (s), 156.32 (s; C=O), 156.53 (s; C=O). — MS (70 eV): m/z (%) = 340 (11, $\text{M}^+ + 1$), 339 (47, M^+), 324 (18), 234 (16), 205 (5), 192 (8), 178 (4), 163 (6), 149 (6), 119 (10), 105 (7), 100 (20), 91 (10), 83 (6), 73 [100, $\text{Si}(\text{CH}_3)_3^+$], 66 (12), 59 (7).

$\text{C}_{18}\text{H}_{21}\text{N}_3\text{O}_2\text{Si}$ (339.5) Calcd. C 63.69 H 6.24 N 12.38
Found C 63.84 H 6.61 N 12.10

2-Chloronorbornadiene (1b**):** From 6.30 g (49.8 mmol) of **1b** and 9.15 g (52.2 mmol) of PTAD in 10 ml of CH_2Cl_2 were obtained the cycloadducts **2b**, **3b**, **4b**, and **5b** after 48 h reaction time and silica gel chromatography using methylene chloride as eluent.

2-Chloro-N-phenyl-3,4-diazatricyclo[4.2.1.0^{2,5}]non-7-ene-3,4-dicarboximide (3b**):** was isolated as first fraction, 300 mg (2%), colorless needles, m.p. 202–203°C (ethanol). — IR (KBr): 3000 cm⁻¹, 1790, 1730, 1600, 1500, 1405, 1320, 1240, 1140, 1025, 970, 780, 755, 720, 690. — ¹H NMR (90 MHz, CDCl_3): δ = AB system ($\delta_A = 2.10$, $\delta_B = 2.65$, $J_{A,B} = 10.2$ Hz; 2 H, 9-H), 3.38 (m; 2 H, 1-H and 6-H), 4.32 (t, $J_{1,5} = J_{5,9a} = 1.6$ Hz; 1 H, 5-H), 6.25 (m; 2 H, 7-H and 8-H), 7.50 (m; 5 H, C_6H_5). — ¹³C NMR (22.6 MHz, CDCl_3): δ = 44.42 (t; C-9), 45.46 (d), 51.92 (d), 74.89 (d), 92.73 (s; C-2), 125.59 (d), 128.81 (d), 129.33 (d), 131.40 (s), 134.94 (d), 137.27 (d), 156.84 (s), 161.14 (s).

$\text{C}_{15}\text{H}_{12}\text{ClN}_3\text{O}_2$ (301.7) Calcd. C 59.71 H 4.01 N 13.93
Found C 59.47 H 4.17 N 13.77

5-Chloro-N-phenyl-2,3-diazabicyclo[4.2.1]nona-4,7-diene-2,3-dicarboximide (5b**):** was isolated as second fraction, 300 mg (2%), colorless plates, m.p. 153–154°C (ethanol). — IR (KBr): 3080 cm⁻¹, 2990, 2930, 2860, 1775, 1720, 1655, 1590, 1500, 1490, 1410, 1305, 1275, 1145, 1065, 830, 755, 735. — ¹H NMR (90 MHz, CDCl_3): δ = 2.15 (d, $J_{9a,9a} = 13.2$ Hz; 1 H, 9s-H), 2.43 (dt, $J_{9a,9a} = 13.2$ Hz, $J_{1,9a} = J_{6,9a} = 6.3$ Hz; 1 H, 9a-H), 3.58 (br. d; 1 H, 6-H), 5.38 (dd, $J_{1,9} = 6.3$ Hz, $J_{1,8} = 2.7$ Hz; 1 H, 1-H), AB system ($\delta_A = 6.20$, $\delta_B = 6.24$, $J_{A,B} = 5.4$ Hz, $J_{6,7} = 3.0$ Hz, $J_{1,8} = 2.7$ Hz; 2 H, 7-H and 8-H), 6.98 (d, $J_{4,6} = 1.5$ Hz; 1 H, 4-H), 7.48 (m; 5 H, C_6H_5). — ¹³C NMR (22.6 MHz, CDCl_3): δ = 39.45 (t; C-9), 51.40 (d), 60.17 (d), 119.32 (d), 122.32 (s), 125.39 (d), 128.29 (d), 129.11 (d), 130.11 (d), 131.14 (s), 136.51 (d), 147.89 (s), 150.01 (s).

$\text{C}_{15}\text{H}_{12}\text{ClN}_3\text{O}_2$ (301.7) Calcd. C 59.71 H 4.01 N 13.93
Found C 59.71 H 4.20 N 13.76

7-Chloro-N-phenyl-4,5-diazatricyclo[4.3.0.0^{2,5}]non-8-ene-4,5-dicarboximide (4b**):** was isolated as third fraction, 300 mg (2%), colorless prisms, m.p. 214–215°C (ethanol). — IR (KBr): 3090 cm⁻¹, 3040, 2990, 2940, 1770, 1710, 1590, 1500, 1400, 1290, 1260, 1230, 1125, 1095, 820, 735, 700, 690. — ¹H NMR (90 MHz, CDCl_3): δ = 1.67 (ddd, $J_{2x,2n} = 12.3$ Hz, $J_{3,2n} = 5.4$ Hz, $J_{2n,1} = 1.5$ Hz; 1 H, 2n-H), 2.23 (dd, $J_{2n,2x} = 12.3$ Hz, $J_{2x,1} = 5.4$ Hz; 1 H, 2x-H), 3.20 (br. s; 1 H, 1-H), 4.55 (m; 2 H, 3-H and 6-H), 6.00 (dd, $J_{8,9} = 6.0$ Hz, $J_{1,8} = 1.4$ Hz; 1 H, 8-H), 6.45 (dd, $J_{8,9} = 6.0$ Hz, $J_{1,9} = 3.6$ Hz; 1 H, 9-H), 7.45 (m; 5 H, C_6H_5). — ¹³C NMR (22.6 MHz, CDCl_3): δ = 37.31 (t; C-2), 43.44 (d), 62.15 (d), 78.74 (s), 84.14 (d), 125.85 (d), 128.45 (d), 129.21 (d), 130.58 (d), 131.70 (s), 138.67 (d), 156.78 (s), 157.49 (s).

$\text{C}_{15}\text{H}_{12}\text{ClN}_3\text{O}_2$ (301.7) Calcd. C 59.71 H 4.01 N 13.93
Found C 59.86 H 3.88 N 13.80

2-Chloro-N-phenyl-8,9-diazatetracyclo[4.3.0.0^{2,4,0^{3,7}]nonane-8,9-dicarboximide (2b**):}** was isolated as fourth fraction, 3.00 g (20%), colorless prisms, m.p. 220–221°C (ethanol). — IR (KBr): 3060 cm⁻¹, 2940, 1765, 1700, 1490, 1410, 1330, 1270, 1220, 1130, 865, 815, 770, 700, 650. — ¹H NMR (90 MHz, CDCl_3): δ = AB system ($\delta_A = 1.86$, $\delta_B = 2.11$, $J_{A,B} = 12.3$ Hz; 2 H, 5-H), 2.07 (m; 2 H, 3-H and

4-H), 2.53 (m; 1H, 6-H), 4.62 (ddm, $J_{6,7} = 2.7$ Hz, $J_{3,7} = 1.5$ Hz; 1H, 7-H), 4.68 (d, $J_{1,6} = 2.7$ Hz; 1H, 1-H), 7.46 (m; 5H, C_6H_5). — ^{13}C NMR (22.6 MHz, $CDCl_3$): $\delta = 24.16, 26.23, 28.84, 43.04, 44.93, 66.24, 70.30, 125.79, 128.52, 129.31, 131.68$.

$C_{15}H_{12}ClN_3O_2$ (301.7) Calcd. C 59.71 H 4.01 N 13.93
Found C 59.42 H 4.16 N 13.81

2-Norbornadienecarbonitrile (1c): From 6.42 g (54.9 mmol) of **1c** and 14.4 g (82.3 mmol) of PTAD in 100 ml of CH_2Cl_2 were obtained the cycloadducts **2c** and **5c** after 8 d reaction time and silica gel chromatography using methylene chloride as eluent.

5-Cyano-*N*-phenyl-2,3-diazabicyclo[4.2.1]nona-4,7-diene-2,3-dicarboximide (5c) was obtained as first fraction, 170 mg (5%), m.p. 88–90°C (ethanol). — IR (KBr): 3060 cm^{-1} , 2200, 1780, 1720, 1625, 1500, 1415, 1310, 1150, 1050, 1020, 935, 830, 750, 690, 645. — 1H NMR (400 MHz, $CDCl_3$): $\delta = 2.11$ (d, $J_{9s,9a} = 13.0$ Hz; 1H, 9s-H), 2.54 (ddd, $J_{9s,9a} = 13.0$ Hz, $J_{1,9a} = 6.4$ Hz, $J_{6,9a} = 6.7$ Hz; 1H, 9a-H), 3.63 (dd, $J_{6,9a} = 6.7$ Hz, $J_{2,6} = 2.9$ Hz; 1H, 6-H), 5.51 (dd, $J_{1,9a} = 6.4$ Hz, $J_{1,8} = 2.8$ Hz; 1H, 1-H), 6.20 (dd, $J_{7,8} = 5.4$ Hz, $J_{1,8} = 2.8$ Hz; 1H, 8-H), 6.29 (dd, $J_{7,8} = 5.4$ Hz, $J_{6,7} = 2.9$ Hz; 1H, 7-H), 7.49 (m; 6H, 5-H and C_6H_5). — ^{13}C NMR (100 MHz, $CDCl_3$): $\delta = 40.67$ (t; C-9), 45.12 (d), 59.16 (d), 96.26 (s), 118.54 (s), 125.44 (d), 128.78 (d), 129.29 (d), 129.62 (d), 130.59 (s), 131.40 (d), 138.72 (d), 147.12 (s), 148.31 (s). — MS (70 eV): m/z (%) = 325 (100, M^+), 206 (21, $M^+ - PhNCO$), 175 (11, PTAD $^+$), 174 (81), 149 (12), 147 (20), 146 (14), 132 (26), 131 (15), 119 (42, PhNCO $^+$), 118 (13), 105 (16), 104 (38), 103 (12), 92 (17), 91 (49), 90 (15), 77 (26), 66 (17), 65 (27), 64 (12), 59 (14), 51 (10).

$C_{17}H_{15}N_3O_4$ (325.2) Calcd. C 62.79 H 4.61 N 12.92
Found C 62.80 H 4.62 N 13.02

2-Methoxycarbonyl-*N*-phenyl-8,9-diazatetracyclo[4.3.0.0^{2,4}.0^{3,7}]nonane-8,9-dicarboximide (2d) was isolated as second fraction, 2.20 g (40%), colorless powder, m.p. 167–168°C (methylene chloride/ether). — IR (KBr): 3475 cm^{-1} , 3075, 3070, 3020, 3010, 2980, 2960, 2940, 2880, 1770, 1730, 1720, 1600, 1590, 1500, 1455, 1435, 1415, 1335, 1300, 1285, 1260, 1230, 1215, 1195, 1160, 1130, 1120, 1095, 1070, 970, 870, 825, 770, 760, 750, 700, 650, 645. — 1H NMR (400 MHz, $CDCl_3$): $\delta = 1.96$ (d, $J_{5s,5a} = 12.0$ Hz; 1H, 5-H), 2.03 (dt, $\delta_B = 2.05$, $J_{5s,5a} = 12.0$ Hz, $J_{4,5} = J_{5,6} = 1.4$ Hz; 1H, 5-H), 2.35 (ddt, $J_{3,4} = 5.1$ Hz, $J_{4,6} = 1.4$ Hz, $J_{4,5} = J_{4,5a} = 1.2$ Hz; 1H, 4-H), 2.42 (m; 1H, 6-H), 2.44 (ddd, $J_{3,4} = 5.1$ Hz, $J_{3,7} = 2.1$ Hz, $J = 0.7$ Hz; 1H, 3-H), 3.76 (s; 3H, CH_3), 4.71 (t, $J_{3,7} = J_{6,7} = 2.3$ Hz; 1H, 7-H), 4.90 (d, $J_{1,6} = 2.5$ Hz; 1H, 1-H), 7.45 (m; 5H, C_6H_5). — ^{13}C NMR (22.6 MHz, $CDCl_3$): $\delta = 26.22, 27.94, 28.49, 29.46, 42.82$ (d), 51.98 (d), 65.49 (d), 66.27 (d), 125.24, 128.06 (s), 128.91, 131.24, 157.10 (s; C=O), 157.40 (s; C=O), 169.87 (s; COO). — MS (70 eV): m/z (%) = 327 (3), 326 (18), 325 (100, M^+), 226 (9, $M^+ - CH_3CO_2$), 206 (3), 175 (5, PTAD $^+$), 174 (28), 149 (12), 147 (17), 135 (8), 132 (10), 119 (39, PhNCO $^+$), 105 (15), 104 (12), 91 (68), 77 (19), 66 (16), 65 (20), 59 (24).

$C_{17}H_{15}N_3O_4$ (325.2) Calcd. C 62.79 H 4.61 N 12.92
Found C 63.05 H 4.60 N 13.01

CAS Registry Numbers

1a: 16205-92-8 / **1b:** 2294-41-9 / **1c:** 39863-20-2 / **1d:** 3604-36-2 / **2a:** 106567-08-2 / **2b:** 82204-55-5 / **2c:** 106567-10-6 / **2d:** 106567-12-8 / **3b:** 106624-48-0 / **4a:** 106567-06-0 / **4'a:** 106567-07-1 / **4b:** 82204-57-7 / **5b:** 82204-58-8 / **5c:** 106567-09-3 / **5d:** 106567-11-7 / HC≡CCN: 1070-71-9 / HC≡CCO₂Me: 922-67-8 / PTAD: 4233-33-4 / norbornadiene: 121-46-0 / 2,6-dichloronorbornene: 59975-41-6 / cyclopentadiene:

- ¹ D. I. Davies, *J. Chem. Soc.* **1960**, 3669.
- ^{2a} C. W. Jefford, P. T. Huy, *Tetrahedron Lett.* **21** (1980) 755. —
^{2b} K. N. Houk, N. G. Rondan, M. N. Paddon-Row, C. W. Jefford, P. T. Huy, P. D. Burrow, K. D. Jordan, *J. Am. Chem. Soc.* **105** (1983) 5563.
- ³ W. Adam, L. A. Arias, O. De Lucchi, *Tetrahedron Lett.* **23** (1982) 399.
- ⁴ W. Adam, O. De Lucchi, *Angew. Chem.* **92** (1980) 815; *Angew. Chem. Int. Ed. Engl.* **19** (1980) 762.
- ⁵ K. Peters, E.-M. Peters, H. G. von Schnering, *Z. Kristal.* **168** (1984) 145.
- ⁶ W. Adam, N. Carballeira, *J. Am. Chem. Soc.* **106** (1984) 2874.
- ⁷ H. D. Verkruisje, L. Braudsma, *Recl. Trav. Chim. Pays-Bas* **105** (1986) 66.
- ⁸ C. D. Buecker, D. Martina, M. F. Neumann, *J. Chem. Res. (S)* **1978**, 78.
- ⁹ S. Murahashi, T. Takizawa, S. Kurioka, S. Maekawa, *Nippon Kagaku Zasshi* **77** (1956) 1689 [*Chem. Abstr.* **53** (1959) 5163f].
- ¹⁰ O. Baumgärtel, G. Szeimies, *Chem. Ber.* **116** (1983) 2180.

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